

所別:藝術學研究所碩士班 乙組科目:美學概論

1. Please choose the compound that *cannot* be prepared by a Williamson ether synthesis, and *explain your choice*. (10 pts)

or or

- 2. Would you expect addition of HCl to the double bond of 3-buten-2-one to follow Markovnikov's rule? **Explain** your answer by a **mechanistic** argument. (10 pts)
- 3. Using the curved-arrow formalism, suggest a mechanism for the following reaction.

 (10 pts)

 H₃C-Li + CH₃CH₂OH
- 4. In the following compounds, determine which one is the most reactive toward electrophilic substitution, and explain your answer. (10 pts)
- 5. In the following pair of ions, determine which ion is more stable. Use resonance structures to explain your answer. (10 pts)
- 6. Why does the diene shown below fail to undergo a Diels-Alder reaction with even the most reactive dienophiles? (10 pts)
- 7. Treatment of Compound 1 at -80 °C with hydrogen bromide gives product A and product B in a 20:80 ratio. After heating at 40 °C, this ratio changes to give mainly product A.

 Propose structures for products A and B. (10 pts)
- 8. Provide the major organic product of the following reaction. (10 pts)

9. Provide the major organic product of the reaction below and a detailed, stepwise mechanism which accounts for its formation. (10 pts)

10. Provide the major organic products (A and B) of the following reactions. (10 pts)