

國立中央大學八十八學年度碩士班研究生入學試題卷

所別: 機械工程研究所 丙組 科目: 熱力學 共 2 頁 第 1 頁

1. Consider an equilibrium mixture of C_2H_6 , O_2 , CO , CO_2 and H_2O . Write down the chemical equations if we take C_2H_6 and CO_2 as independent variables. (10%)
2. For a vessel containing vapor H_2O and liquid H_2O , assume that $T_f = T_g = 100\text{ }^\circ\text{C}$, $p_f = 0.1013\text{ Mpa}$, $p_g = 0.07\text{ Mpa}$: Is the system in equilibrium? If not, will more vapor H_2O be condensed or more liquid H_2O be vaporized, why? [For saturated H_2O at $100\text{ }^\circ\text{C}$: $p_{sat} = 0.1013\text{ Mpa}$, $h_f = 419\text{ kJ/kg}$, $s_f = 1.3071\text{ kJ/(kgK)}$; For superheated H_2O at $100\text{ }^\circ\text{C}$ and 0.07 Mpa : $h = 2680\text{ kJ/kg}$, $s = 7.5341\text{ kJ/(kgK)}$] (10%)
3. Consider the entropy of a substance in its three phases.. (10%)
 - (a) In which phase, the entropy is highest? In which phase, the entropy is lowest?
 - (b) Give an explanation of your answers in part (a).
4. One important method to achieve low temperatures is by means of the Joule-Thomson expansion. (15%)
 - (a) Define the Joule-Thomson coefficient, μ_{JT} .
 - (b) To be able to lower the temperature, should μ_{JT} of the working fluid be positive or negative?
 - (c) Show that the Joule-Thomson coefficient of an ideal gas equals to zero. Therefore, we cannot lower the temperature of an ideal gas by use of the Joule-Thomson expansion.
5. A combined cycle involves a gas power cycle topping a vapor power cycle. Show that the thermal efficiency of a combined gas-steam power plant η_{cc} can be expressed as
$$\eta_{cc} = \eta_g + \eta_s - \eta_g \eta_s,$$
where $\eta_g = W_g/Q_{in}$ and $\eta_s = W_s/Q_{g,out}$ are the thermal efficiencies of the gas and the steam cycles, respectively. (5%)
6. Plate glass at $500\text{ }^\circ\text{C}$ is cooled by passing air over its surfaces such that the convection heat transfer coefficient is $h = 10\text{ W/m}^2\text{-K}$. To prevent cracking, it is known that the temperature gradient must not exceed $15\text{ }^\circ\text{C/mm}$ at any point in the glass during the cooling process. If the thermal conductivity of the glass is 1.4 W/m-K and its surface emissivity is 0.8 , what is the lowest temperature of the air that can be initially used for the cooling? Assume that the temperature of the air equals that of the surroundings. (10%)

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7. A long cylinder of radius 10 cm consists of a nuclear reacting material ($k = 0.5 \text{ W/m-K}$) generating $24,000 \text{ W/m}^3$ uniformly throughout its volume. This rod is encapsulated within another cylinder having an outer radius of 20 cm and a thermal conductivity of 4 W/m-K . The outer surface is surrounded by a fluid at $50 \text{ }^\circ\text{C}$, and the convection coefficient between the surface and the fluid is $20 \text{ W/m}^2\text{-K}$. Find the temperature at the interface between the two cylinders and at the outer surface. (15%)
8. It is popular to apply Dittus-Boelter equation, $Nu = 0.023Re_p^{0.8}Pr^n$, for estimating in-tube heat transfer coefficient. Please explain why the Prandtl number exponent $n = 0.3$ for cooling and $n = 0.4$ for heating of fluid. (10%)
9. You are requested to design a test facility to determine the local heat transfer coefficient of uniform air flow over a flat plate.
 - (a) Sketch a complete schematic diagram of the test facilities. Describe all of the components of the system and what data will be taken. (10%)
 - (b) Write down all data reduction equations for obtaining the local heat transfer coefficient from the original data you measured. (5%)

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