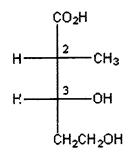
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一、單選題 (答案請填於答案卡,答錯不倒扣,每題 2.5 分,共 70 分)

1. Identify all pairs of diastereomers.

- (A) B and D, A and B, A and D, B and C, C and D
- (B) A and B, A and D, B and C, C and D
- (C) B and D
- (D) B and D, A and C
- (E) A and C
- 2. In the Fischer projection below, what are the configurations of the two asymmetric centers?



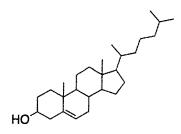
- (A) 2S,3R
- (B) 2R,3S
- (C) 2R,3R
- (D) 2*S*,3*S*
- (E) cannot be determined from structure
- 3. Which of the following is a meso compound?
- (A) cis-1,4-dimethylcyclohexane
- (B) trans-1,4-dimethylcyclohexane
- (C) trans-1,2-dimethylcyclohexane
- (D) trans-1,3-dimethylcyclohexane
- (E) cis-1,3-dimethylcyclohexane
- 4. Keto-enol forms are
- (A) enantiomers
- (B) constitutional isomers
- (C) conformational isomers

- (D) diastereomers
- (E) resonance structures

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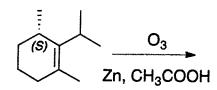
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5. Structure of cholesterol is shown below. How many stereoisomers exist for this molecule?



- (A) 8 (B) 32
- (C) 64
- (D) 128
- (E) 256

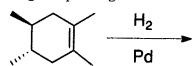
6. In the following ozonolysis reaction, if we start with optically pure S-enantiomer, the product obtained will be



- (A) meso compounds.
- (B) optically pure S-enantiomer.
- (C) racemic mixture.

- (D) diastereomers.
- (E) unequal mixture of R and S enantiomers.

7. Hydrogenation of the following compound generates



- (A) diastereomers.
- (B) identical compounds. (C) enantiomers.
- (D) constitutional isomers.

(E) meso compounds.

8. Which of the following has two equatorial alkyl substituents in its most stable conformation?

- (A) trans-1,3-diethylcyclohexane
- (B) cis-1,2-dimethylcyclohexane
- (C) cis-1,4-diethylcyclohexane
- (D) 1,1-dimethylcyclohexane
- (E) cis-1,3-diethylcyclohexane

9. Which is the correct order of decreasing acidity in the following compounds?

 H_2O

CH₃CH₃

В

 \mathbf{C}

 NH_3

D

CH₂=CH₂ HC≡CH E

(A) E > D > B > A > C

- (B) A > E > C > D > B
- (C) A > E > D > B > C

- (D) A > C > E > D > B
- (E) E > A > C > B > D

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10. Which species is represented by the following distribution of p electrons in the molecular energy diagram?

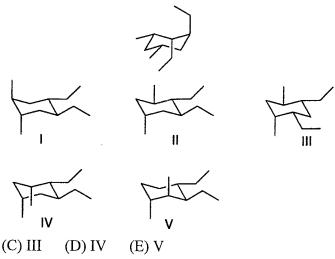
- (A) I (B) II (C) III (D) IV (E) V
- 11. Indicate the compound that has the shortest bond length between the two middle carbon atoms.
- (A) $CH \equiv C-C \equiv CH$ (B) $CH_3CH_2-CH = CH_2$ (C) $CH_2 = CH-C \equiv CH$
- (D) CH₃CH₂-C≡CH (E) CH₂=CH-CH=CH₂
- 12. Which of the following alkyl bromides is likely to undergo rearrangement by a 1,2-methyl shift?
- (A) 2-bromo-3, 3-dimethylpentane
- (B) 3-bromo-3-methylpentane
- (C) 3-bromo-2, 3-dimethylpentane
- (D) benzyl bromide
- (E) 2-bromo-3-ethylpentane
- 13. What is the major product which results when (2R,3S)-2-chloro-3-phenylbutane is treated with sodium methoxide in methanol?
- (A) (S)-3-phenyl-1-butene
- (B) (R)-2-methoxy-2-phenylbutane (C) (E)-2-phenyl-2-butene

(D) (Z)-2-phenyl-2-butene

(A) I

(B) II

- (E) (R)-3-phenyl-1-butene
- 14. Which of the following represents the ring flipped structure of the molecule below?



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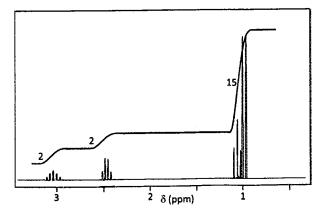
- 15. H-A is an acid with a pK_a of 4.5. Which of the following statements about an aqueous solution of H-A is true?
- (A) At pH = 3.5, the solution contains about 90% A- and 10% H-A.
- (B) At pH = 5.5, the solution contains about 90% A- and 10% H-A.
- (C) At pH = 4.5, the solution contains much more A- than H-A.
- (D) At pH = 4.5, the solution contains much more H-A than. A-.
- (E) At pH = 6.5, the solution contains about 80% A- and 20% H-A.
- 16. Provide the IUPAC name of the compound.

$$CH_3$$
 $H_3C-C-CH_3$
 H_3C^N
 CH_3

- (A) N,N,2-trimethyl-1-propanamine
- (B) N,N-dimethyl-2-butanamine (C) N,N,2-trimethylpropanamine
- (D) N,N,1,1-tetramethylethanamine
- (E) N,N,2-trimethyl-2-propanamine
- 17. Which of the following compound is most acidic?
- (A)
 - 0 0
- (R)
- о о
- (C)
- (D)
- O O CF₂
- 0

(E)

18. Which of the following compounds has the ¹H-NMR spectrum shown below (all ¹H signals are shown)?



- (A)
- _o__
- (B)
- N
- (C)
- O N
- (D)



(E)



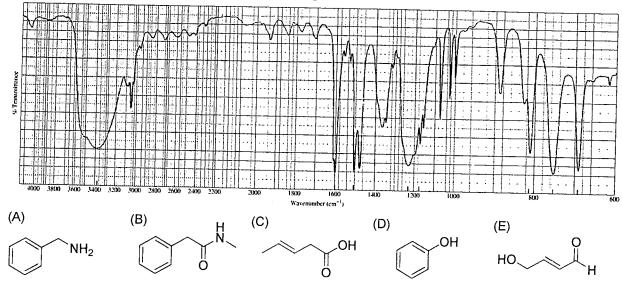
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19. What is the major product for the following reaction?

$$(A) \qquad (B) \qquad (C) \qquad (D) \qquad (E)$$

$$OH \qquad OH \qquad OH$$

20. Which of the following compounds has the IR spectrum shown below?



21. Which of the following carbohydrate structure is the cyclized form of the hexose in Fischer projection?

$$(A) \qquad (B) \qquad (C) \qquad (D) \qquad (E)$$

$$(A) \qquad (B) \qquad (C) \qquad (D) \qquad (E)$$

$$(A) \qquad (B) \qquad (C) \qquad (D) \qquad (E)$$

$$(A) \qquad (B) \qquad (C) \qquad (D) \qquad (D)$$

$$(C) \qquad (D) \qquad (D) \qquad (D)$$

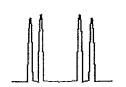
$$(D) \qquad (D)$$

注意:背面有試題

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The signal peaks in the figure are entirely and the only ¹H-NMR signals of indicated protons of the shown compound. What is the R group?



(A)

(B)

(C)

(D)

(E)

R = H

 $R = CH_2OH$

R = Ph

23. What is the major product of the reaction shown below?

- 1. NaOCH₂CH₃
- 2. CH₃(CH₂)₂Br

(A)

(B)

(C)

(D)

(E)

24. What is the major product of the reaction shown below?

(B)

(D)

(E)

(C)

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25. Which set of reactions can lead to the shown compound as major product?

(D) Br
$$\frac{1. \text{ CH}_2\text{CH}_2, \text{ PdCl}_2}{2. \text{ O}_3, -78^{\circ}\text{C}}$$

 $\frac{2. \text{ O}_3, -78^{\circ}\text{C}}{3. \text{ (CH}_3)_2\text{S}}$

26. Which set of conditions lead to the shown reaction?

(A) O_3 , -78°C then $(CH_3)_2S$

(C) CH₂CH₂, PdCl₂

(D)
$$Ph_3P$$
 PPh_3 + n -BuLi

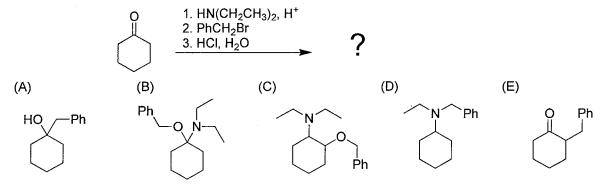
(E)
$$R_3P$$
 CI
 PR_3 Ph
 $R = I$

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27. What is the major product of the reaction shown below?

28. What is the major product of the reaction shown below?

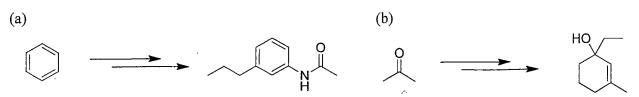


二、問答題 (共30分)

1. Show detailed step-by-step mechanism for the following \mathbf{Hg}^{2+} catalyzed hydration for a terminal alkyne. (5%)

$$CH_3CH_2C\equiv CH + H_2O \xrightarrow{H_2SO_4} CH_3CH_2 \stackrel{O}{\leftarrow} CH_3CH_2$$

- 2. Nomenclature. Give the IUPAC name or draw the structure (5%)
 - (a) H (b)(5*E*)-4-ethynylocta-5,7-dien-2-yn-1-ol
- 3. Provide reagents (no more than 5 steps) for the following synthesis: (10%)



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- 4. To synthesize 3-hydroxypentanal from propanal, acetaldehyde and lithium diisopropylamide, what is the suitable order to add the above reagents to the reaction flask? (5%)
- 5. Predict the major product for the following reaction? (5%)